Combustion synthesis of near-infrared reflective brown pigments based on iron-doped lanthanum aluminate

<u>Robert Ianoș*a</u>, Eliza Munteana, Radu Lazăua, Roxana Băbuțăa, Alina Moacăa, Cornelia Păcurariua, Anamaria Dabici^b, Roxana Istratie^b

a. Politehnica University of Timişoara, Faculty of Industrial Chemistry and Environmental Engineering, P-ţa Victoriei no. 2, Timişoara 300006, Romania, *E-mail address: robert_ianos@yahoo.com, b. National Institute for Research & Development in Electrochemistry and Condensed Matter, Department of Condensed Matter, Plautius Andronescu Street, No. 1, Timişoara 300224, Romania



Fig. 5. XRD patterns (A) of the samples. BET surface area and crystallite size (B) as a function of iron (x) content.



 $6 \operatorname{La}(NO_3)_3 + 10 \operatorname{C}_2H_5NO_2 + 6(1-x) \operatorname{Al}(NO_3)_3 + 6x \operatorname{Fe}(NO_3)_3 + 15 \operatorname{CH}_4N_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 55 \operatorname{H}_2O + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 35 \operatorname{CO}_2 + 38 \operatorname{N}_2O = 6 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 36 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 36 \operatorname{CO}_2 + 36 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 36 \operatorname{La}\operatorname{Al}_{1-x}\operatorname{Fe}_xO_3 + 36 \operatorname{La}\operatorname{Al}_{1-x}O_3 + 36 \operatorname{La}\operatorname{Al}_{1-x}O_2 + 36 \operatorname$



Fig. 1. Still images captured during the evolution of a typical combustion reaction (x = 0.75, sample LAF 0.75).



Table 1. Total solar reflection (TSR) and CIEL*a*b* color evaluation of $LaAl_{1-x}Fe_xO_3$ pigments and coatings.

Sample	Pigment				Acrylic coating (30 wt. % pigment)				Wet / dry coating
	TSR (%)	L *	a*	b *	TSR (%)	L *	a*	b *	thickness (µm)
LAF 0	85.3	98.1	0.3	3.3	87.5	96.5	-0.4	3.1	300 / 94
LAF 0.25	57.3	77.4	1.1	27.1	49.4	69.7	2.7	28.5	300 / 91
LAF 0.5	47.6	65.0	5.0	33.6	38.5	58.1	6.7	32.6	300 / 101
LAF 0.75	42.7	59.0	10.2	39.5	35.8	53.7	12.2	35.1	300 / 99













Fig. 4. Images of the coatings applied onto Al sheets (60x60 mm).

Characterization techniques:

• Thermal imaging: FLIR T 640, 15 frames/second.

XRD: Rigaku Ultima IV, Cu_{κα}.

- **TG-DSC**: Netzsch STA 449, 10 C⁹/min heating rate, air atmosphere, alumina crucibles.
- BET: Micromeritics ASAP 2020, samples degassed at 300 C^o and 5 μ mHg for 12 hours.
- TEM: FEI Titan G2 80–200.
- **DRS**: Perkin Elmer Lambda 950, D65 illuminant, observer's angle 10, CIEL*a*b*, TSR.
- Coating thickness: Byko-test 1500 film gauge.



Fig. 8. DRS spectra of the pigments (A) and coatings (C). Colour position in the CIEL*a*b* colour space (B).



NIR-reflective brown pigments based on $LaAl_{1-x}Fe_xO_3$ (x = 0 - 0.75) were prepared by solution combustion synthesis, without any additional thermal treatment.

As the iron amount increases $(0 \rightarrow 0.75)$ combustion temperature decreases $(1712 \rightarrow 1017 \text{ }^{\circ}\text{C})$, leading to a decrease of crystallite size $(57 \rightarrow 22 \text{ nm})$ and an increase of BET surface area $(3.0 \rightarrow 15.6 \text{ m}^2/\text{g})$.

In terms of colour, the brown shade of the pigments gets more intense for larger iron amounts but the total solar reflectance decreases from 85.3 % to 42.7 %.

The combustion-synthesized pigments were successfully tested for the preparation of water-based acrylic paints of different brown shades having NIR-reflective capabilities.

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